

# coefficient of isothermal compressibility using the equation of state for an ideal gas

none

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## Abstract

Generated by the [Physics Derivation Graph](https://notendur.hi.is/hj/EE2/HD1lausn.pdf). <https://notendur.hi.is/hj/EE2/HD1lausn.pdf>

Eq. 1 is an initial equation.

$$\kappa_T = \frac{-1}{V} \left( \frac{\partial V}{\partial P} \right)_T \quad (1)$$

Eq. 2 is an initial equation.

$$PV = nRT \quad (2)$$

Divide both sides of Eq. 2 by  $P$ ; yields Eq. 3.

$$V = \frac{nRT}{P} \quad (3)$$

Substitute LHS of Eq. 3 into Eq. 1; yields Eq. 4.

$$\kappa_T = \frac{-1}{V} \left( \frac{\partial}{\partial P} \left( \frac{nRT}{P} \right) \right)_T \quad (4)$$

Simplify Eq. 4; yields Eq. 5.

$$\kappa_T = \frac{-nRT}{V} \left( \frac{\partial}{\partial P} \left( \frac{1}{P} \right) \right)_T \quad (5)$$

Simplify Eq. 5; yields Eq. 6.

$$\kappa_T = \frac{-nRT}{V} \left( \frac{-1}{P^2} \right) \quad (6)$$

Substitute LHS of Eq. 2 into Eq. 6; yields Eq. 7.

$$\kappa_T = \frac{-PV}{V} \left( \frac{-1}{P^2} \right) \quad (7)$$

Simplify Eq. 7; yields Eq. 8.

$$\kappa_T = \frac{1}{P} \quad (8)$$

Eq. 8 is one of the final equations.

## References